Synthesis of Some 2-Amino-4-phenyl-3*H*-1,5-benzodiazepines by Reaction of Aromatic Diamines and Acetylenic Imidates

Paul C. Unangst

Department of Chemistry, Warner-Lambert Company, 2800 Plymouth Road
Ann Arbor, Michigan 48105
Received May 1, 1981

The reaction of several aromatic 1,2-diamines with conjugated acetylenic imidate salts is described. The use of hexamethylphosphoric triamide (HMPA) as the solvent allows preparation of 2-amino-4-phenyl-3*H*-1,5-benzodiazepines in a one-step reaction.

J. Heterocyclic Chem., 18, 1257 (1981).

The 2-amino-4-phenyl-3H-1,5-benzodiazepines are of interest as analogues of the 1,4-benzodiazepine anti-anxiety agents (Valium[®], Librium[®]), as well as certain neuroleptic agents (1,2). A number of 1,5-benzodiazepines are under clinical investigation for a variety of therapeutic applications (3).

Many 1,5-benzodiazepines have been prepared by con-

densation of an aromatic 1,2-diamine and β -dicarbonyl compounds, α,β -olefinic carbonyl compounds (4-6), keto-keteneacetals (7), keto-ketenethioacetals (8), keto-dithioacetic acids (1,9), or dichloromethyleniminium salts (10). Less common is the reaction of a 1,2-diamine and an ethynylcarbonyl compound (11-13).

We previously reported (14) the synthesis of the 2-ami-

$$R_{1}-C \equiv C-C \searrow_{NH_{2}}^{O}$$

$$I = -C$$

$$R_{1}-C \equiv C-C \searrow_{NH_{2}}^{OE1}$$

$$R_{1}-C \equiv C-C \searrow_{NH_{2}}^{OE1}$$

$$R_{2}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{1}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{2}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{1}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{2}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{3}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{1}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{2}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{2}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{2}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{3}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{1}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{2}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{2}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{3}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{2}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{3}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{2}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{3}-C \equiv C-C \searrow_{NH_{2}}^{NH_{2}}$$

$$R_{4}-C \equiv C-C \searrow_{NH_{2}}^{NH_{$$

Scheme I

Compound

Substituent	a	b	c	d	e	f	g
R,	2-ClC₀H₄	4-ClC ₆ H ₄	C_6H_5	C ₆ H ₅	C_6H_5	C_6H_5	C_6H_5
R₂	Н	Н	Н	Cl	NO_2	Н	Н
R_s	NH_2	NH ₂	NH ₂	NH_2	NH ₂	N	N NCH3

Table I
2-Amino-4-phenyl-3H-1,5-benzodiazepines

Compound (a)	Formula	% Yield	Anal. Calcd. (Found) %		
(a)		(crude)	С	H	N
			66.79	4.48	15.58
4a	$C_{15}H_{12}CIN_3$	51 (b)	(66.87	4.99	15.64)
	20 20 0		66.79	4.48	15.58
4b	$C_{15}H_{12}CIN_3$	41 (c)	(66.53	4.42	15.45)
			64.27	4.32	19.99
4e	$C_{15}H_{12}N_4O_2$	36 (c)	(64.20	4.57	19.80)
			58.84	4.28	13.72
5b	C ₁₅ H ₁₂ ClN ₃ •HCl	41 (c)	(58.68	4.49	13.45)
			66.30	5.19	15.46
5c	C,sH,3N, HCl	56 (c)	(66.15	5.20	15.35)
	10 10 0		58.84	4.28	13.72
5d	C ₁₅ H ₁₂ ClN ₃ •HCl	34 (c)	(58.73	4.31	13.61)
	10 12 0		79.17	6.98	13.85
6f	$C_{20}H_{21}N_3$	49	(79.01	7.04	13.81)
	20 21 0		75.44	6.96	17.60
6 g	$\mathrm{C_{20}H_{22}N_4}$	68	(75.26	7.21	17.55)

(a) Substituents are given in Scheme I. (b) Prepared from amidine 3a. See reference (14). (c) Yield is that of the free base product calculated with the inclusion of one equivalent of HMPA.

no-1,5-benzodiazepine 4a (Scheme I) by acid-catalyzed cyclization in ethanol of the stable acetylenic amidine 3a. The amidine resulted from condensation of 2-aminoaniline (o-phenylenediamine) and the free imidate base derived from acetylenic imidate salt 2a. We have now found that additional compounds analogous to 4a can be prepared in one step from the imidates 2a-c without prior neutralization of the imidate salts or isolation of the intermediate amidines 3a-e. The key to this process appears to be the use of hexamethylphosphoric triamide ("HMPA") as the reaction solvent (17).

The acetylenic amides la-c were prepared by literature (14-16) procedures and were converted to the imidate salts 2a-c by treatment with triethyloxonium tetrafluoroborate ("Meerwein's Reagent") in methylene chloride. When the salts 2b,c were combined with an aromatic 1,2-diamine in HMPA and heated on the steam bath, benzodiazepines 4b-e were obtained.

In most instances, the crude product was obtained as a 1:1 complex with HMPA. The ability of HMPA to catalyze nucleophilic addition reactions, as well as form stable addition complexes with amines is well known (18). The HMPA was removed by recrystallization and high-vacuum drying of the crystals or by converstion of the free base benzodiazepines to the hydrochloride salts (e.g., 5b-d). Hydrochloride salt 5c was transaminated by heating with excess piperidine or 1-methylpiperazine to form benzodia-

zepines **6f,g**, two compounds prepared in the literature (1,19) by displacement of a benzodiazepine methylthioimino ether. The benzodiazepines obtained are listed in Table I.

The assignment of the R_2 substituent of $\mathbf{4e}$ and $\mathbf{5d}$ (Table I) to the #8 position in the aromatic ring (rather than the #7 position) presupposes the probable formation of amidines $\mathbf{3d}$ and $\mathbf{3e}$ as transient intermediates in the formation of $\mathbf{4e}$ and $\mathbf{5d}$. In addition, a comparison of the ultraviolet spectra (20) of the known 1,5-benzodiazepin-2-ones $\mathbf{7}$ and $\mathbf{8}$ as model compounds, indicates the spectrum of nitrobenzodiazepine $\mathbf{4e}$ is clearly quite similar to that of

7, while substantially dissimilar to that of the alternate isomer 8.

We have described the preparation of several 2-amino-1,5-benzodiazepines by a simple, one-step reaction of aromatic 1,2-diamines with acetylenic imidate salts.

EXPERIMENTAL

Melting points were determined in a Mel-Temp capillary apparatus and are uncorrected. Nmr spectra were recorded on a Varian EM-390 spectrometer at 90 MHz, or a Perkin-Elmer R-12B spectrometer at 60 MHz, with tetramethylsilane as an internal standard in both cases. Infrared spectra were recorded on a Beckman DK-I spectrophotometer or a Digilab FTS-14 pulsed Fourier-transform spectrophotometer. All uv spectra were determined in 95% ethanol on a Cary Model 118 spectrophotometer.

Ethyl 3-Phenyl-2-propynimidate Tetrafluoroborate (2c).

A suspension of 32.0 g (0.22 mole) of 3-phenyl-2-propynamide 1c (15) in 450 ml of dry dichloromethane was cooled in ice and treated over 2.0 hours with a solution of 42.1 g (0.22 mole) of triethyloxonium tetrafluoroborate. The ice bath was removed, and the mixture was stirred at room temperature for 3 days. The imidate salt product 2c was filtered, and the filtrate was condensed to 75 ml and treated with ether to precipitate additional product. After being washed several times with fresh ether, the combined crops yielded white solid of mp 117-120° (49 g, 85% yield). A sample recrystallized from dichloromethane had mp 121-123°; ir (chloroform): ν 3300, 3140 (NH), 2220 (C \equiv C), 1676 (C=N), 1100 (broad, BF4) cm⁻¹; nmr (DMSO-d₆): δ 1.41 (t, 3H, J = 7.5 Hz, CH₂CH₃, 4.67 (q, 2H, J = 7.5 Hz, CH₂-CH₃), 7.25-7.87 (m, 5H, ArH), 10.50 (broad s, 2H, NH₂).

Anal. Calcd. for $C_{11}H_{12}BF_4NO$: C, 50.62; H, 4.63; N, 5.37; F, 29.11. Found: C, 50.48; H, 4.61; N, 5.44; F, 29.08.

Ethyl 3-(4-Chlorophenyl)-2-propynimidate Tetrafluoroborate (2b).

The procedure described above for 2c was employed in the preparation of 2b except that after addition of triethyloxonium tetrafluoroborate solution, the reaction mixture was heated to reflux for 30 minutes (became one phase), filtered hot, and allowed to cool slowly. From 15.0 g (0.083 mole) of 3-(4-chlorophenyl-2-propynamide 1b (15,16) there was obtained 17.2 g (70% yield) of imidate salt 2b, white solid of mp $139-143^\circ$; ir (Nujol): ν 3320, 3180 (NH), 2230 (C \equiv C), 1690 (C=N), 1050 (broad, BF₄) cm⁻¹; nmr (DMSO-d₆): δ 1.47 (t, 3H, J = 7.0 Hz, CH₂CH₃), 4.70 (q, 2H, J = 7.0 Hz, CH₂CH₃), 7.20-8.20 (m, 4H, ArH), 11.12 (broad s, 2H, NH₂). This material was used without further purification for the preparation of benzodiazepine 4b.

4-(4-Chlorophenyl)-3H-1,5-benzodiazepin-2-amine (4b).

A mixture of 5.0~g (0.017 mole) of the imidate salt 2b and 1.8~g (0.017 mole) of 1,2-benzenediamine (o-phenylenediamine) in 50 ml of hexamethylphosphoric triamide (HMPA) was stirred and heated on the steam bath for 3 hours. The cooled reaction mixture was added to 300 ml of 1% aqueous sodium hydroxide cooled in ice. After stirring for 30 minutes, the semi-solid product was extracted with chloroform (3 × 125 ml). The combined organic layers were back-washed with water (2 × 150 ml), dried (sodium sulfate), and evaporated under reduced pressure. The residue was treated with cold hexane (~350 ml) to form a yellow solid 4b, which was filtered and washed several times with fresh cold hexane. Recrystallization of the crude product from benzene/hexane formed yellow cubes of mp 192-196° (3.1 g, 41% yield as •1 HMPA). The ir and nmr spectra of the recrystallized material indicated the incorporation of one equivalent of HMPA in the crystal. A second recrystallization of this material from ethanol/water followed by high-vacuum drying at 78° gave an analytical sample of 4b, yellow solid of mp 198-200°, containing no solvent of crystallization; ir (Nujol): v 3500, 3320 (NH), 1653 (C=N), 1590, 1306, 1002, 754 cm⁻¹; nmr (DMSO-d₆): δ 3.25 (s, 2H, CH₂), 6.95 (broad s, 2H, NH₂), 7.05-8.33 (m, 7H, ArH); uv: λ max nm (ϵ) 214 (32,880), 254 (33,480), 345 (6920).

2-Amino-4-(4-chlorophenyl)-5H-1,5-benzodiazepinium Chloride (5b).

The hydrochloride 5b was obtained by dissolving a sample of the crude HMPA-complex of 4b in a minumum of chloroform, followed by ice cooling and the addition of excess gaseous hydrogen chloride to the solution. A volume of ether twice that of the original chloroform solution was added to precipitate the product. The crude hydrochloride was

filtered, washed with fresh ether, and recrystallized twice from ethanol, to yield red solid **5b**, mp 305° dec, containing no incorporated solvent; ir (Nujol): ν 3238, 3088 (NH), 1642 (C=N), 1582, 1548, 1208, 1075, 758 cm⁻¹; nmr (DMSO-d₆): δ 3.82 (s, \sim 0.2-0.3 H, C H_2 tautomer), 4.80 (s, \sim 0.8 H, CH = tautomer), 6.68-7.71 (m, 8H, ArH), 8.38 (broad s, 1H, NH), 9.06 (broad s, 2H, N H_2), 10.53 (broad s, 1H, NH); uv: λ max nm (ϵ) 206 (31,500), 264 (29,000), 335 (3500).

The procedures described above for 4b and 5b were also employed to prepare 4e, 5c and 5d.

2-Amino-4-phenyl-5H-1,5-benzodiazepinium Chloride (5c).

There was obtained 54.7 g (56% yield) of the HMPA-complex of 4c from reaction of 61.2 g (0.24 mole) of imidate 2c and 25.8 g (0.24 mole) of 1,2-benzenediamine in 125 ml of HMPA. Conversion of 4c to the hydrochloride 5c yielded a red solid of mp 285° dec; ir (Nujol): ν 3270, 3120 (NH), 1662 (C=N), 1585, 1555 (broad), 1228, 758 cm⁻¹ cm⁻¹; nmr (DMSOd6): δ 3.87 (s, \sim 0.2-0.3H, CH₂ tautomer) 4.84 (s, \sim 0.8-0.9H, CH = tautomer), 6.65-7.75 (m, 9H, ArH), 8.39 (broad s, 1H, NH), 9.07 (broad s, 2H, NH₂), 10.58 (broad s, 1H, NH); uv: λ max nm (ϵ) 204 (29,800), 264 (26,900), 335 (2500).

2-Amino-8-chloro-4-phenyl-5H-1,5-benzodiazepinium Chloride (5d).

Reaction of 16.7 g (0.064 mole) of imidate 2c and 9.2 g (0.064 mole) of 4-chloro-1,2-benzenediamine in 30 ml of HMPA yielded 9.7 g (34% yield) of (4d) as the HMPA-complex. Conversion of 4d to the hydrochloride 5d gave an orange solid of mp 295° dec; ir (Nujol): ν 3390, 3175 (NH), 1662 (C=N), 1600, 1558 (broad), 1232, 1108, 762 cm⁻¹; nmr (DMSO-d₆): δ 3.97 (s, ~0.2-0.3H, CH₂ tautomer), 4.92 (s, ~0.8-0.9H, CH, CH = tautomer), 6.80-7.92 (m, 8H, ArH), 8.53 (broad s, 1H, NH), 9.28 (broad s, 2H, NH₂), 10.92 (broad s, 1H, NH); uv: λ max nm (ϵ) 216 (25,000), 264 (29,400), 240 (3250).

8-Nitro-4-pheny-3H-1,5-benzodiazepin-2-amine (4e).

The procedure described in the prepartion of **4b** was employed to prepare 3.1 g (36% yield) of crude nitro-benzodiazepine **4e** from 8.0 g (0.03 mole) of imidate **2c** and 4.7 g (0.03 mole) of 4-nitro-1,2-benzenediamine in 20 ml of HMPA. When the initial reaction mixture was added to 400 ml of 1% aqueous sodium hydroxide and 125 ml of chloroform, a yellow precipitate formed. Filtration and recrystallization of a sample from acetone yielded **4e** as yellow needles of mp 280° dec, containing no incorporated solvent after high-vacuum drying at 78°; ir (Nujol): ν 3460, 3350 (NH), 1673 (C=N), 1558, 1508, 1330 (broad, NO₂), 1077, 760 cm⁻¹; nmr (DMSO-d₆): δ 3.35 (s, 2H, CH_2), 7.60 (broad s, 2H, NH_2), 7.18-8.58 (m, 8H, ArH); uv: λ max nm (ϵ) 206 (30,320), 252 (17,720), 298 (16,200), 350 (15,280).

2-Phenyl-4-(1-piperidinyl)-3H-1,5-benzodiazepine (6f).

A suspension of 8.2 g (0.03 mole) of the hydrochloride 5c in 80 ml of absolute ethanol was cooled in ice and treated dropwise over 30 minutes with a solution of 12.9 g (15.0 ml, 0.15 mole) of piperidine in 30 ml of ethanol. The ice bath was removed, and the mixture was stirred at reflux under nitrogen for 67 hours. The cooled reaction mixture was filtered and evaporated under vacuum.. The residue was partitioned between 300 ml of 0.3N sodium hydroxide solution and 100 ml of chloroform. The aqueous layer was separated and washed with fresh chloroform (2 imes 100 ml). The combined organic layers were back-washed with water (1 imes 150 ml), dried (sodium sulfate) and evaporated to an oil. Trituration with cold hexane vielded a yellow solid (4.5 g, 49% yeild). Recrystallization of a sample from ethanol/water yielded 6f as yellow needles of mp 110-113°; literature (1,19) mp 115-118°, 119-121°; ir (Nujol): v 1595, 1570 (broad), 1428, 1202, 959, 762 cm $^{-1}$; nmr (deuteriochloroform): δ 1.12-1.71 (m, 6H, piperidine CH₂), 3.22-3.82 (m, 6H, piperidine CH₂ and diazepine CH₂), 6.87-8.28 (m, 9H, ArH); uv: \(\lambda\) max nm (\(\epsi\)) 210 (24,880), 261 (33,120), 350 (4440).

2-(4-Methyl-1-piperazinyl)-4-phenyl-3H-1,5-benzodiazepine (6g).

The procedure described in the preparation of 6f was employed to prepare 4.0 g (68% yield) of crude piperazinylbenzodiazepine 6g from

5.0 g (0.018 mole) of hydrochloride $\mathbf{5c}$ and $\mathbf{4.5}$ g (5.0 ml, 0.045 mole) of 1-methylpiperazine in 75 ml of absolute ethanol. Recrystallization of a sample from ethanol/ether yielded $\mathbf{6g}$ as yellow crystals of mp 151-154°, lit (1,19) mp 155-157°, 158°; ir (Nujol): ν 1601, 1573 (broad), 1412, 1210, 975, 782 cm⁻¹; nmr (deuteriochloroform): δ 2.08-2.54 (m, 7H, piperazinyl CH₂ and CH₃) 3.38 (s, 2H, diazepine CH₂), 3.48-3.86 (t, 4H, J = 5.0 Hz, piperazinyl CH₂), 6.91-8.18 (m, 9H, ArH).

Acknowledgements.

We are indebted to Mrs. U. Zeek (Warner-Lambert-New Jersey) and Mr. C. E. Childs (Warner-Lambert-Michigan) and staffs for microanalyses and to Dr. C. Greenough (New Jersey) and Dr. F. A. MacKellar (Michigan) and staffs for spectral data.

REFERENCES AND NOTES

- (1) C. R. Ellefson, C. M. Woo, A. Miller and J. R. Kehr, J. Med. Chem., 21, 952 (1978).
 - (2) L. H. Sternbach, ibid., 22, 1 (1979).
 - (3) J. G. Berger and L. C. Iorio, Ann. Rep. Med. Chem., 15, 22 (1980).
- (4) Z. F. Solomko and A. N. Kost, Chem. Heterocyclic Compounds, 11, 1231 (1975).
 - (5) C. A. Archer and L.H. Sternbach, Chem. Rev., 68, 747 (1968).
- (6) D. Lloyd and H. P. Cleghorn, in "Advances in Heterocyclic Chem.", Vol. 17, A. R. Katritzky and A. J. Boulton, eds., Academic Press,

New York, NY, 1974, p. 27.

- (7) H. D. Stachel, Chem. Ber., 95, 2172 (1962).
- (8) A. Ushirogochi, Y. Tominaga, Y. Matsuda and G. Kobayashi, Heterocycles. 14. 7 (1980).
- (9) D. Nardi, R. Pennini and A. Tajana, J. Heterocyclic Chem., 12, 825 (1975).
- (10) H. Böhme and H. G. Viehe, in "Advances in Organic Chemistry", Vol. 9, E. C. Taylor, ed., John Wiley and Sons, Inc. New York, NY, 1976, p. 343.
 - (11) W. Ried and R. Teubner, Ann. Chem., 741 (1978).
- (12) Y. S. Andreichikov, S. G. Pitirimova, S. P. Tendryakova, R. F. Saraeva and T. N. Tokmakova, Zh. Org. Khim., 14, 169 (1978).
- (13) S. P. Korshunov, V. M. Kazantseva, L. A. Vopilina, V. S. Pisareva and N. V. Utekhina, ibid., 9, 1287 (1973).
- (14) P. C. Unangst and P. L. Southwick, J. Heterocyclic Chem., 10, 399 (1973).
 - (15) H. Adkins and G. Whitman, J. Am. Chem. Soc., 64, 150 (1942).
- (16) J. Schmitt, French Patent 1,305,340 (1962); Chem. Abstr., 58, 7874c (1963).
- (17) Caution: Hexamethylphosphoric triamide is a suspected carcinogen, and should be handled with appropriate precautions.
 - (18) H. Normant, Angew. Chem., Int. Ed. Engl., 6, 1046 (1967).
- (19) A. Tajana, R. Pennini and D. Nardi, Farmaco, Ed. Sci., 35, 181 (1980).
- (20) T. S. Chmilenko, Z. F. Solomko and A. N. Kost, Chem. Heterocyclic Compounds, 13, 423 (1977).